

# Alterations in Acid-Base Balance

Nursing Pathophysiology (N – 4)

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## Acid-Base Balances

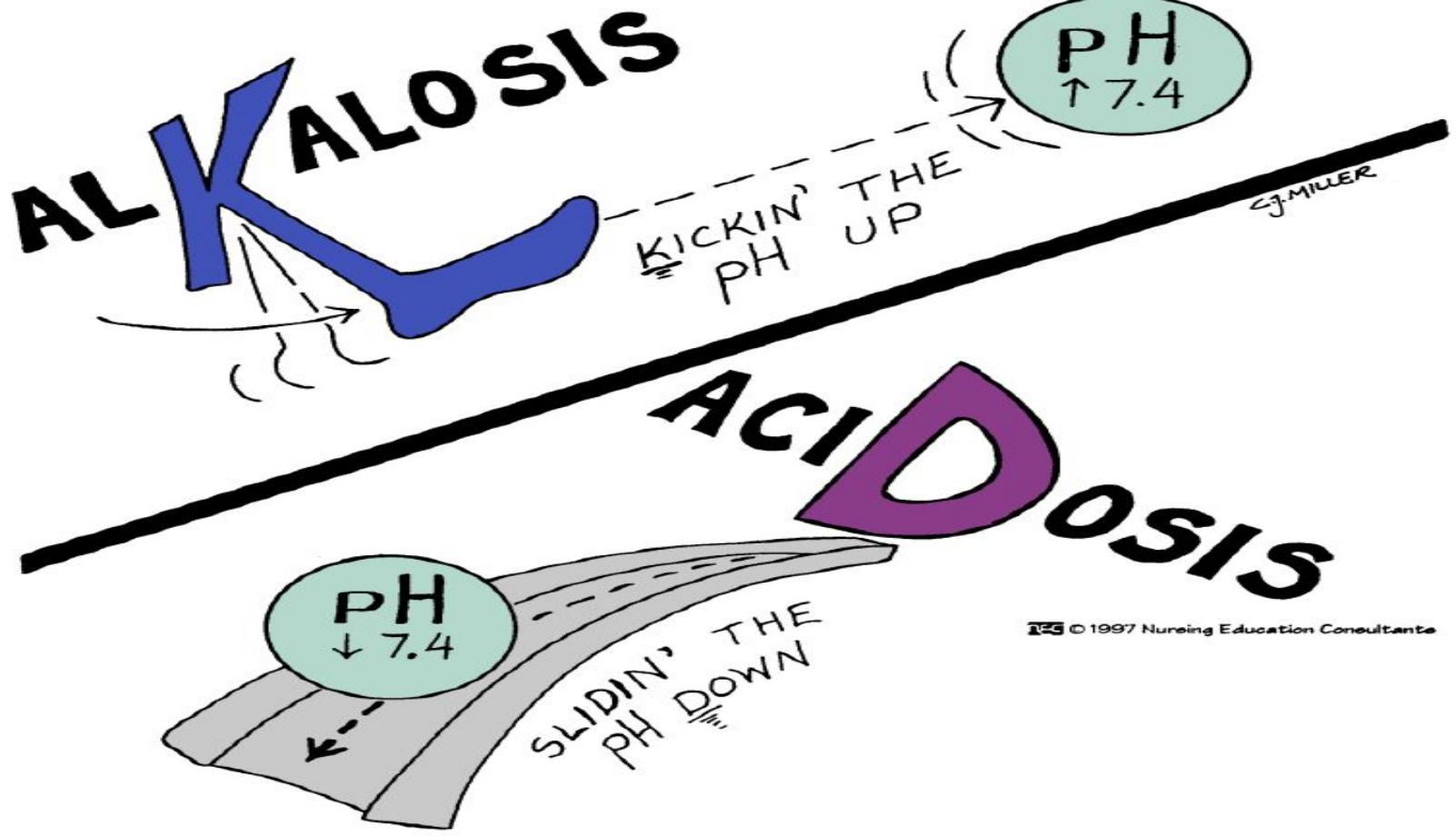
- PH 7.35-7.45
- Acids/Bases – hydrogen ion
- Buffer Systems – promote balance
  - Bicarbonate/ Carbonic acid
  - Alkalosis – above 7.45
  - Acidosis – below 7.35

# Acid-Base Disorders

pathologic changes in **carbon dioxide partial pressure ( $\text{Pco}_2$ )** or **serum bicarbonate ( $\text{HCO}_3^-$ )** that typically produce abnormal arterial pH values.

- **Acidemia** is serum pH < 7.35
- **Alkalemia** is serum pH > 7.45
- **Acidosis** refers to physiologic processes that cause acid accumulation or alkali loss.
- **Alkalosis** refers to physiologic processes that cause alkali accumulation or acid loss.

## ACIDOSIS - ALKALOSIS



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# Sources of Acids and Bicarbonate Ions

## Acids

- Carbon dioxide
- Fatty acids and ketoacids
- Anaerobic – lactic acid and ketoacids
- Impaired cells

## Bicarbonate

- Breakdown of carbonic acid, intestinal absorption, pancreatic production, movement of cellular bicarbonate in ECF and kidney reabsorption of bicarbonate

# Acid Base Regulatory Mechanisms

- Chemical Acid-Base Control
  - Bicarbonate (ECF & ICF) and phosphate (ICF)
- Respiratory Acid Base Control
  - Carbon dioxide
- Renal Acid Base Control
  - Bicarbonate, acids, ammonium



Basic

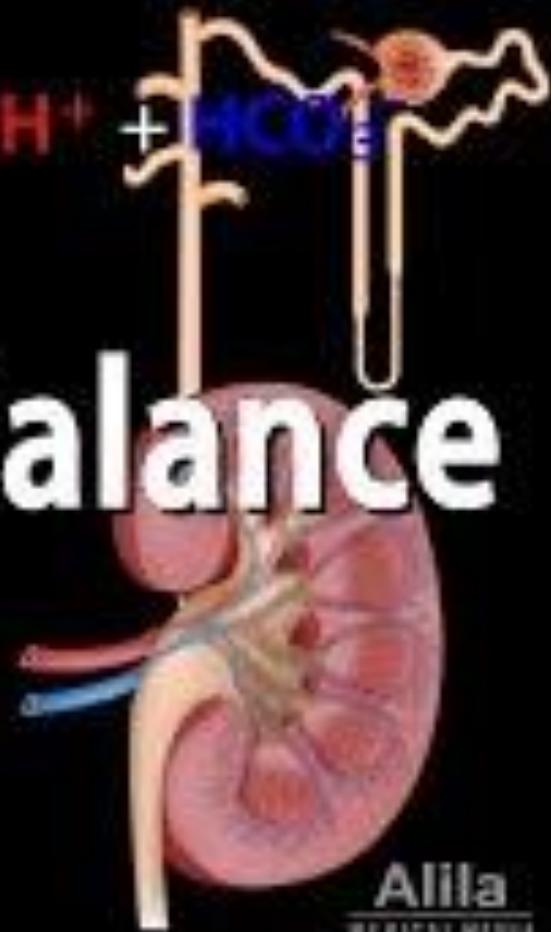
Neutral

Acidic

# Acid-Base Balance

Acid

Base

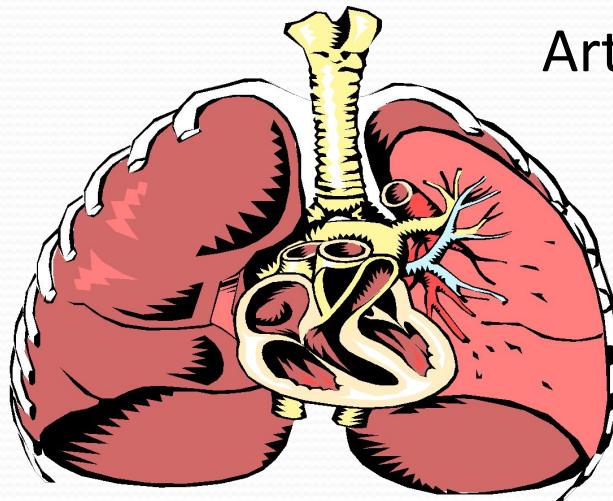


Alila  
MEDICAL MEDIA

# Biological Compensation

## Respiratory

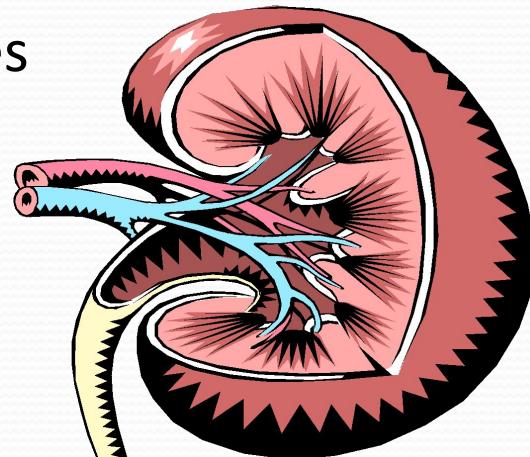
- Metabolic disorders – Diabetes, acute renal failure
- carbon dioxide retained or excreted



## Renal

- In lung disorders – COPD
- Formation of acids or bicarbonate reabsorbed or excreted

Arterial Blood Gases  
(ABGs)



# Classification of Acid-Base Disorders

ACIDOSIS	ALKALOSIS
<p><b><u>Respiratory acidosis</u></b> is <math>\text{Pco}_2 &gt; 40</math> mm Hg (hypercapnia). Cause is</p> <ul style="list-style-type: none"><li>Decrease in minute ventilation (hypoventilation)</li></ul>	<p><b><u>Respiratory alkalosis</u></b> is <math>\text{Pco}_2 &lt; 38</math> mm Hg (hypocapnia). Cause is</p> <ul style="list-style-type: none"><li>Increase in minute ventilation (hyperventilation)</li></ul>
<p><b><u>Metabolic acidosis</u></b> is serum <math>\text{HCO}_3^- &lt; 24</math> mEq/L (&lt; 24 mmol/L). Causes are</p> <ul style="list-style-type: none"><li>Increased acid production</li><li>Acid ingestion</li><li>Decreased renal acid excretion</li><li>Gastrointestinal or renal <math>\text{HCO}_3^-</math> loss</li></ul>	<p><b><u>Metabolic alkalosis</u></b> is serum <math>\text{HCO}_3^- &gt; 28</math> mEq/L (&gt; 28 mmol/L). Causes are</p> <ul style="list-style-type: none"><li>Acid loss</li><li><math>\text{HCO}_3^-</math> retention</li></ul>

# Acidosis

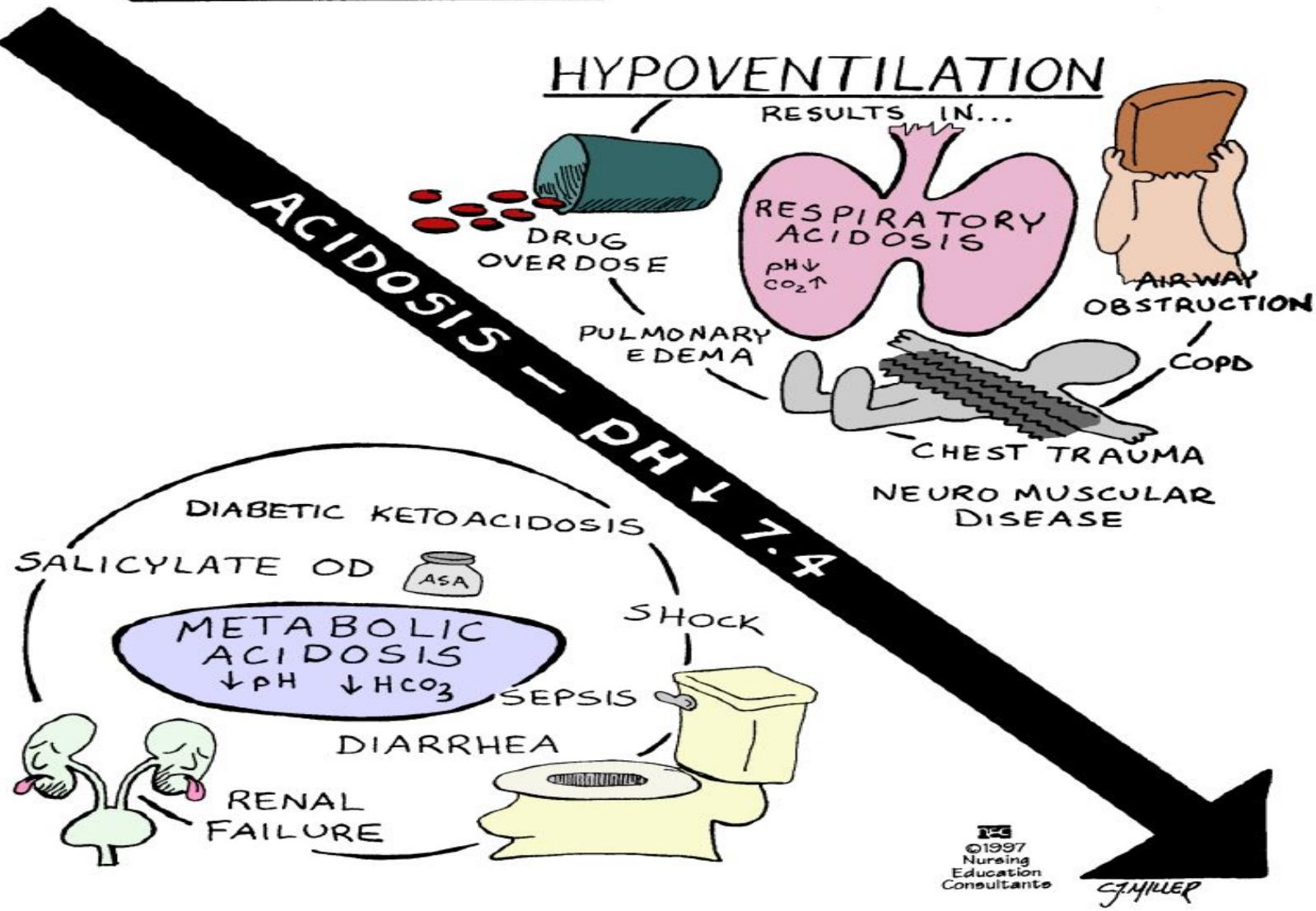
## RESPIRATORY

- Respiratory Depression
- Inadequate Chest Expansions
- Airway Obstruction
- Reduced Alveolar-capillary diffusion

## METABOLIC

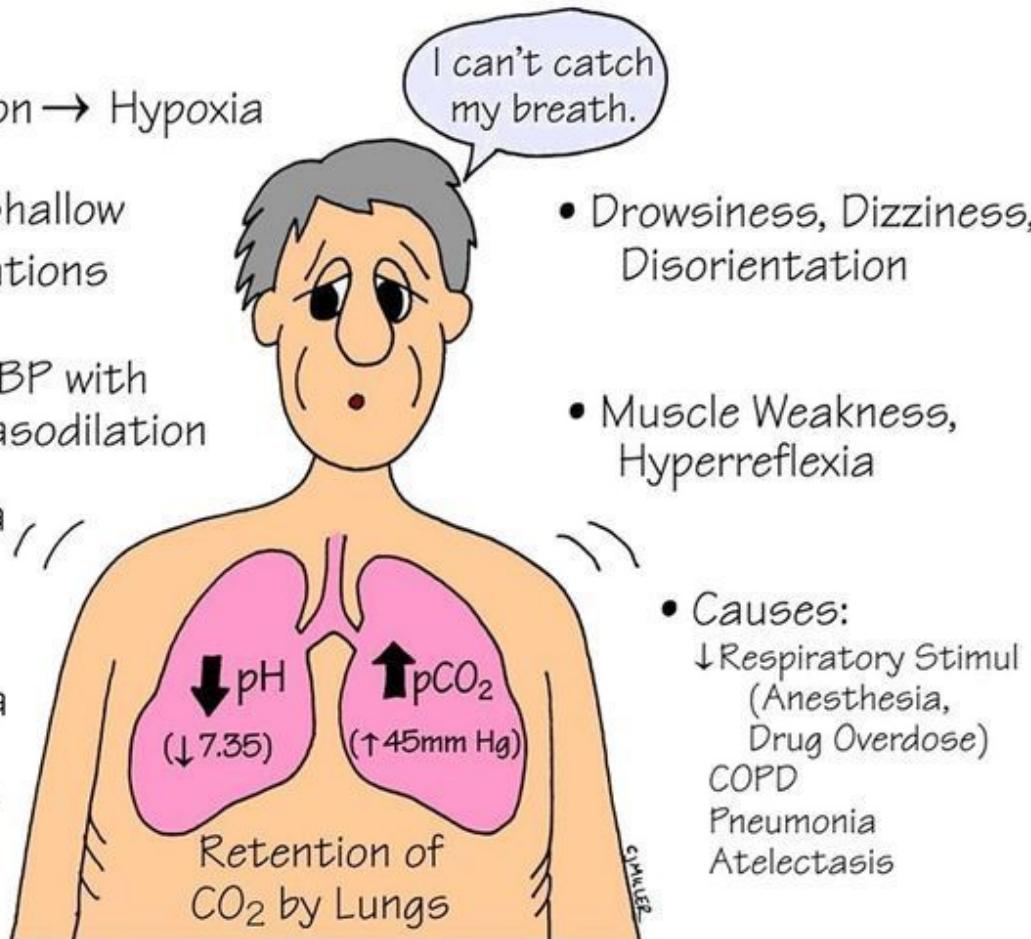
- Overproduction of *Hydrogen* Ions
- Underelimination of *Hydrogen* Ions
- Underproduction of *Bicarbonate* Ions
- Overelimination of *Bicarbonate* Ions

## ..... IMBALANCES:



# RESPIRATORY ACIDOSIS

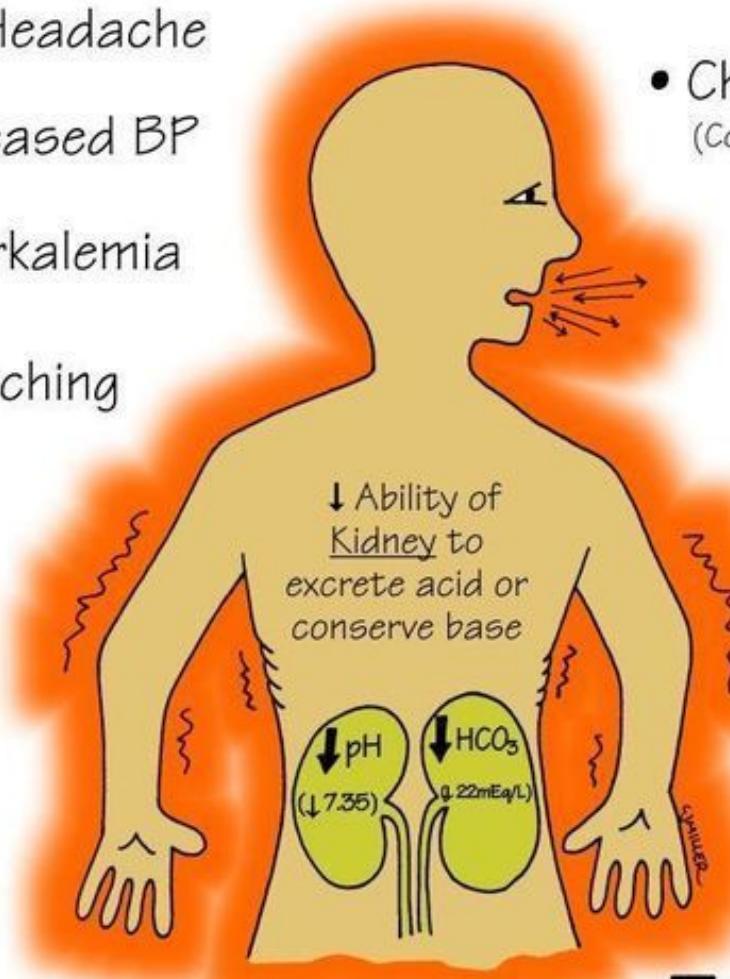
- Hypoventilation → Hypoxia
  - Rapid, Shallow Respirations
  - ↓ BP with Vasodilation
  - Dyspnea
  - Headache
  - Hyperkalemia
  - Dysrhythmias ( $\uparrow K$ )



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# METABOLIC ACIDOSIS

- Headache
- Decreased BP
- Hyperkalemia
- Muscle Twitching
- Warm, Flushed Skin  
(Vasodilation)
- Nausea, Vomiting Diarrhea



- Changes in LOC  
(Confusion, ↑ drowsiness)
- Kussmaul Respirations  
(Compensatory Hyperventilation)
- Causes:  
DKA  
Severe Diarrhea  
Renal Failure  
Shock

# Assessment of Acidosis

- History – age, cause, diet, medications, illness
- Physical Assessment
  - Lethargic, confusion, coma
  - Muscle weakness, deep tendon reflexes, flaccid paralysis; skin in metabolic - warm, dry, pink (due to vasodilation); skin in respiratory – pale to cyanotic.
  - Heart rate ↑, then in severe cases, heart rate ↓ ,  
BP, monitor vs, O2 sat, EKG
  - Mental status – confused, uncooperative
  - Metabolic acidosis – low bicarbonate; Respiratory acidosis – elevated carbonic acid (CO<sub>2</sub>)

# Treatment for Acidosis

- Metabolic
  - Hydration
  - Treat cause – diabetic Ketoacidosis – insulin; antidiarrheal for diarrhea
  - Dialysis – renal failure
  - Monitor VS, EKG
  - Assess for complications
  
- Respiratory
  - Oxygen, bronchial dilators, dry pulmonary secretions, breathing exercise, postural drainage
  - Monitor oxygen sat levels, VS, EKG
  - Assess for complications

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# CAUSES OF ACID-BASE.....

## HYPERVENTILATION

RESULTS IN...



INITIAL STAGES OF PULMONARY EMBOLI



FEVER



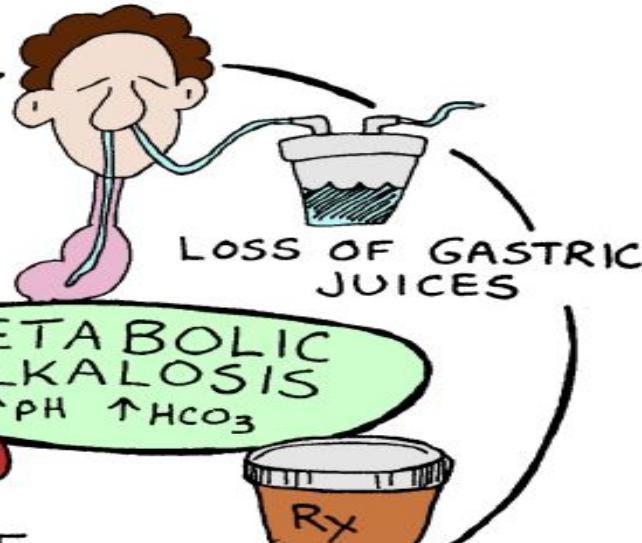
PREGNANCY



ALKALOSIS - PH ↑ 7.4



OVERUSE OF ANTACIDS



METABOLIC ALKALOSIS  
↑ pH ↑ HCO<sub>3</sub>

POTASSIUM-WASTING DIURETICS  
(↑ Loss of H<sup>+</sup>)

CJ MILLER

# Alkalosis

## METABOLIC

- Base Excess
- Acid Deficits

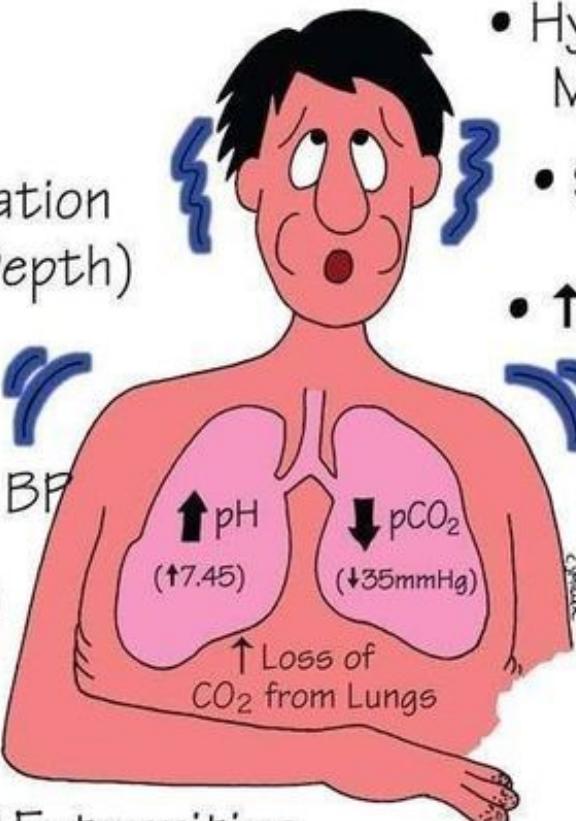
## RESPIRATORY

- Loss of carbonic acid in hyperventilation
- Anxiety, fear or improper settings on mechanical ventilators
- Hyperventilation direct stimulation of CNS – fever, metabolic acidosis, drugs - salicylates

# METABOLIC ALKALOSIS

- Confusion
  - Dysrhythmias  
(Tachycardia from  $\downarrow K^+$ )
  - Compensatory Hypoventilation
  - Causes:
    - $\uparrow HCO_3$  (Antacids, admin of sodium bicarbonate)
    - $\downarrow H^+$  (NG Suctioning, Prolonged Vomiting, Hypercortisolism)
- 
- Dizzy,  $\uparrow$  Irritability
  - Nausea, Vomiting, Diarrhea
  - $\uparrow$  Anxiety, Seizures
  - Tremors, Muscle Cramps, Tingling of Fingers & Toes ( $\downarrow$  serum  $Ca^{++}$ )

# RESPIRATORY ALKALOSIS

- Hyperventilation  
(↑Rate & Depth)
  - Tachycardia
  - ↓or Normal BP
  - Hypokalemia
  - Numbness & Tingling of Extremities
- 
- Hyper Reflexes & Muscle Cramping
  - Seizures
  - ↑ Anxiety, ↑ Irritability
  - Causes:
    - Hyperventilation (Anxiety, PE, Fear)
    - Mechanical Ventilation



# Assessment of Alkalosis

- Physical Assessment
  - CNS- dizziness, agitation confusion, hyperreflexia, paresthesia, Chvostek's and Trousseau's signs.
  - Cramps, twitches, charley horses, deep tendon reflexes hyperactive, tetany, weak muscles, poor hand grasp
  - Heart rate ↑, pulse thready, BP ↓
- ↑Rate and depth of respirations
- Laboratory
  - Metabolic – elevated bicarbonate
  - Respiratory – low bicarbonate and carbonic acid

# Treatment for Alkalosis

- Treat the cause
  - Correct electrolyte imbalances; remove if excess or administer if low
  - Hydration
  - Antiemetic for upper GI distress
  - Monitor IV fluids, VS, ABG's, I&O, oxygen, respiratory and cardiac (EKG)
  - Assess for complications

# ACID BASE MNEMONIC **(ROME)**

**R**

**R**espiratory

**O**

**O**pposite

pH ↑ PCO<sub>2</sub> ↓ Alkalosis

pH ↓ PCO<sub>2</sub> ↑ Acidosis

**M**

**M**etabolic

**E**

**E**qual

pH ↑ HCO<sub>3</sub> ↑ Alkalosis

pH ↓ HCO<sub>3</sub> ↓ Acidosis



# Interpretation of Acid Base Disturbance

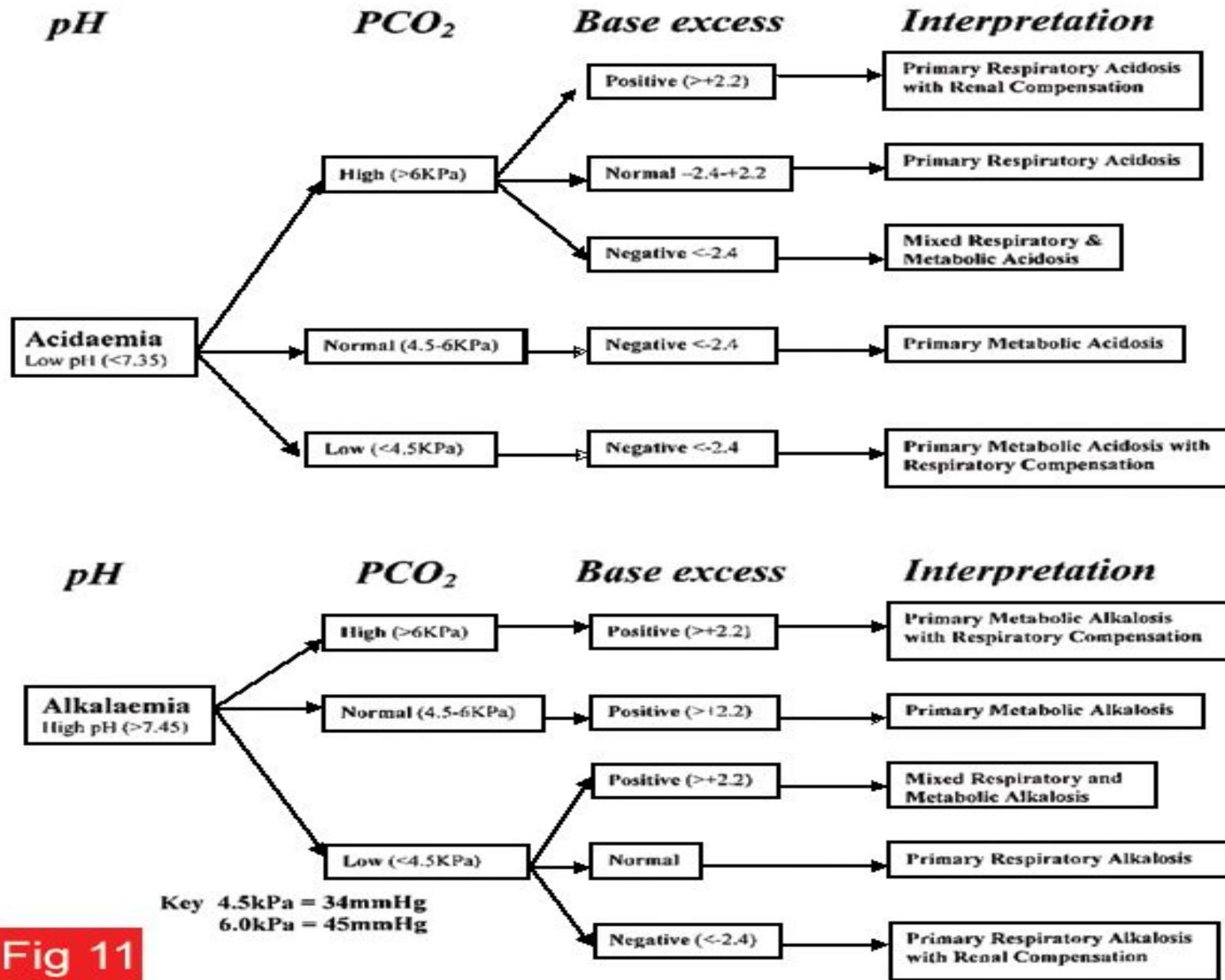


Fig 11

## Exercise:

1. pH 7.51, pCO<sub>2</sub> 40, HCO<sub>3</sub><sup>-</sup> 31: **B**
- a. Normal
  - b. Uncompensated metabolic alkalosis
  - c. Partially compensated respiratory acidosis
  - d. Uncompensated respiratory alkalosis

## Exercise:

2. pH 7.33, pCO<sub>2</sub> 29, HCO<sub>3</sub><sup>-</sup> 16:

- a. Uncompensated respiratory alkalosis
- a. Uncompensated metabolic acidosis
- b. Partially compensated respiratory acidosis
- c. Partially compensated metabolic acidosis

D

## Exercise:

3. pH 7.40, pCO<sub>2</sub> 40, HCO<sub>3</sub><sup>-</sup> 24:

A

- a. Normal
- b. Uncompensated metabolic acidosis
- c. Partially compensated respiratory acidosis
- d. Partially compensated metabolic acidosis

## Exercise: **C**

4. pH 7.12, pCO<sub>2</sub> 60, HCO<sub>3</sub><sup>-</sup> 29:

- a. Uncompensated metabolic acidosis
- b. Uncompensated respiratory acidosis
- c. Partially compensated respiratory acidosis
- d. Partially compensated metabolic acidosis